



# Preparation of polyfunctional pyridines by a palladium(0)-catalyzed cross-coupling of functionalized aryl Grignard reagents

Véronique Bonnet,<sup>a</sup> Florence Mongin,<sup>b</sup> Francois Trécourt,<sup>b</sup> Guy Quéguiner<sup>b</sup> and Paul Knochel<sup>a,\*</sup>

<sup>a</sup>Ludwig-Maximilians-Universität München, Institut für Organische Chemie, Butenandtstr. 5-13, Haus F, 81377 München, Germany

<sup>b</sup>Laboratoire de Chimie Organique Fine et Hétérocyclique de l'IRCOF associé au CNRS, INSA de Rouen, BP 08, 76131 Mont-Saint-Aignan, France

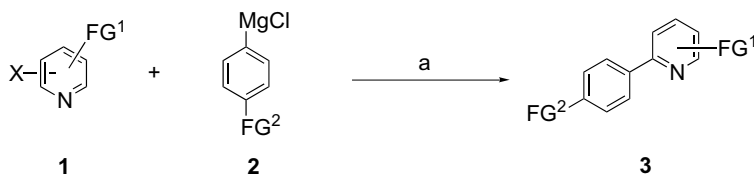
Received 21 June 2001

**Abstract**—Difunctionalized pyridines can be prepared by a Pd(0)-catalyzed cross-coupling of functionalized arylmagnesium compounds with chloro- or bromopyridines at temperatures as low as  $-40^{\circ}\text{C}$ . An addition–elimination mechanism involving a palladate intermediate is proposed. © 2001 Elsevier Science Ltd. All rights reserved.

The preparation of polyfunctional heterocycles is an important synthetic goal because of the multiple applications of these molecules.<sup>1</sup> Transition metal cross-couplings have proven to be an important method for preparing a number of complex heterocycles.<sup>2</sup> The major limitation of this approach is the moderate reactivity of typical organometallics used, such couplings as organozincs, organoboronic acids or organotin derivatives.<sup>3</sup> More reactive organometallics suffer from a moderate functional group tolerance. Recently, we have reported a new preparation of polyfunctional aryl- and heteroarylmagnesium compounds via a halogen–magnesium exchange reaction.<sup>4</sup> Herein, we wish to report remarkably mild reaction conditions allowing a palladium-catalyzed cross-coupling of functionalized pyridyl halides of type **1** with functionalized arylmagnesium reagents of type **2** leading to polyfunctional pyridines of type **3** (see Scheme 1 and Table 1). Already

several Stille, Suzuki and Negishi cross-coupling reactions have been reported with chloropyridines and chloroquinolines.<sup>5</sup> In most cases, these reactions proceed in refluxing THF. These conditions are not suitable for functionalized Grignard compounds of type **2** bearing an ester or a cyano group. More appropriate would be a related nickel-catalyzed reaction using pyridyl bromides,<sup>6</sup> however the toxicity of nickel salts led us to explore an alternative route.

We found that by treating phenylmagnesium chloride (**2a**: 1.2 equiv.) with ethyl 6-bromonicotinate (**1a**: 1.0 equiv.) with bis(dibenzylideneacetone)palladium(0); (Pd(dba)<sub>2</sub>: 5–10 mol%) and 1,1'-bis(diphenylphosphino)ferrocene (dppf, 5–10 mol%)<sup>7</sup> in THF at  $-40^{\circ}\text{C}$  for 6 h, a fast cross-coupling reaction is observed leading to the 6-phenyl-substituted nicotinic derivative **3a** in 95%

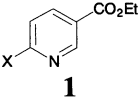
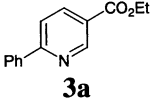
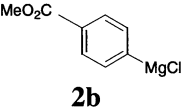
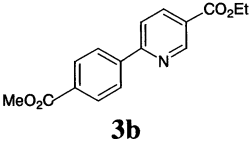
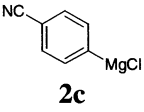
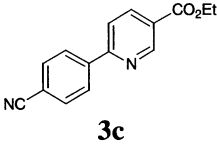
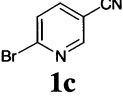
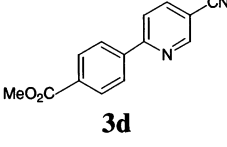
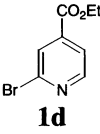
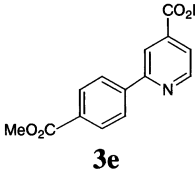
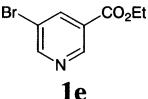
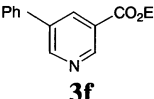
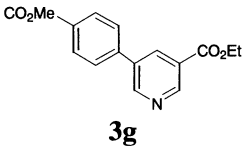
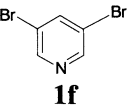
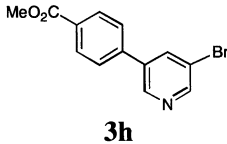


**Scheme 1.** Cross-coupling reactions of functionalized magnesium reagents: X = Cl, Br; FG<sup>1</sup> = CN, CO<sub>2</sub>Et, Br; FG<sup>2</sup> = H; CO<sub>2</sub>Me, CN; (a) Pd(dba)<sub>2</sub> (5–10% mol), dppf or *t*-Bu<sub>3</sub>P (5–10% mol).

**Keywords:** palladium-catalysis; pyridine synthesis; functionalized organomagnesium reagent; heterocycle synthesis.

\* Corresponding author. Fax: +49-(0)89-2180-7680; e-mail: paul.knochel@cup.uni-muenchen.de

**Table 1.** Polyfunctional pyridines **3a–3h** obtained by a palladium(0)-catalyzed cross-coupling of arylmagnesium reagents with chloro- or bromo-pyridines

Entry	Grignard reagent	Pyridyl halide	Conditions (°C, h)	Product of Type 3	Yield (%) <sup>a</sup>
					
1	<b>2a:</b> PhMgCl	<b>1a:</b> X=Br	-40, 6	<b>3a</b>	95
2	<b>2a:</b> PhMgCl	<b>1b:</b> X=Cl	-40, 6		92
3		<b>1b</b>	-40, 6		95
4		<b>1a</b>	-40, 6		87
5	<b>2b</b>		-40, 6		86
6	<b>2b</b>		-20, 4		90
7	<b>2a</b>		0, 18 <sup>b</sup>		73
8	<b>2b</b>	<b>1e</b>	0, 18 <sup>b</sup>		63
9	<b>2b</b>		-5, 18 <sup>b</sup>		62

<sup>a</sup>Yield of analytically pure products<sup>b</sup>This reaction has been performed with *t*-Bu<sub>3</sub>P (10 mol%) and Pd(dba)<sub>2</sub> (10 mol%).

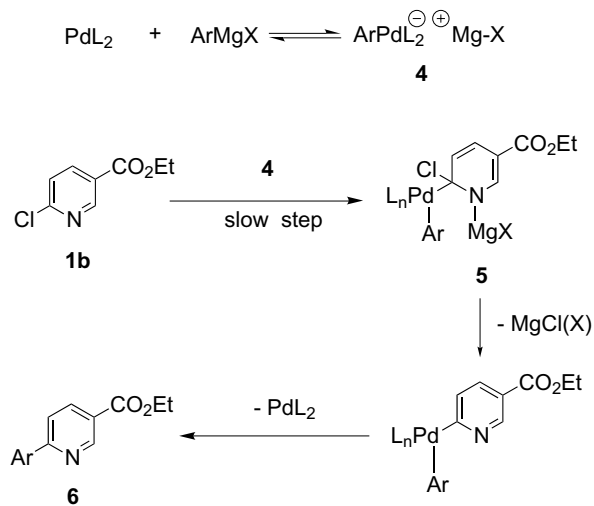
yield. Remarkably, the same smooth reaction is observed with the corresponding chloropyridine **1b** leading to **3a** in 92% yield (entries 1 and 2 of Table 1).

These mild conditions allow the extension of this cross-coupling to functionalized arylmagnesium compounds such as the ester-substituted Grignard reagent (**2b**) and 4-cyanophenylmagnesium chloride (**2c**). These two organometallics are stable up to 0°C. We have tested

the cross-coupling reaction between 2-bromo-4-methylpyridine and phenylmagnesium chloride (**2a**) with the usual catalyst system at -25°C over 18 h. No deprotonation of the methyl group was noticed<sup>8</sup> and we obtained the cross-coupled product quantitatively.<sup>9</sup> The exceptionally fast palladium-catalyzed cross-coupling with 2-halogenopyridines can be explained assuming an addition–elimination mechanism to 2-chloropyridine (**1b**). The high electrophilicity of the pyridine facilitates

the initial addition step to PdLn or more likely of the related ate-species  $\text{ArPdLn}^- \text{MgX}^+$  (**4**) obtained by the reaction of  $\text{PdL}_2$  with the Grignard reagent leading to the stabilized magnesium amide **5**, which after elimination of a chloride and subsequent reductive elimination furnishes the product **6** (Scheme 2).<sup>10</sup> We have observed that the cross-coupling reaction proceeds under such mild conditions only when using an arylmagnesium reagent. The corresponding arylzinc derivative does not react under the same conditions and only gives a slow reaction at higher temperature (0°C) supporting that the oxidative addition may involve the palladate species **4**. Organozinc reagents have less tendency to form ate-species due to the more covalent nature of the C–Zn bond. The importance of the carboxy group in position **3**, which is able to stabilize the excess negative charge of **5** by resonance, can be established since the corresponding 4-substituted carboxy-2-bromopyridine reacts only at –20°C with the Grignard reagent **2b** leading to the cross-coupling product **3e** in 90% yield (entry 6 of Table 1).<sup>11</sup> No reaction occurs in the absence of the palladium(0) catalyst. 3-Halo-substituted pyridines also react but since the extra-stabilization due to the formation of an intermediate of type **5** is no longer possible, higher temperatures are required for the cross-coupling. In these cases it has been advantageous to use *t*-Bu<sub>3</sub>P as ligand<sup>12</sup> (10 mol%). Under these conditions a cross-coupling with phenylmagnesium chloride (**2a**) and 4-carbomethoxyphenylmagnesium chloride (**2b**) can be achieved at 0°C within 18 h in good yields (63–73%; entries 7 and 8).

In summary, we have found very mild reaction conditions allowing the palladium-catalyzed cross-coupling of functionalized arylmagnesium compounds with various chloro- and bromo-pyridines leading to polyfunctional pyridines which are of interest for various pharmaceutical applications.



**Scheme 2.** Possible mechanism for the Pd(0)-catalyzed substitution of 2-chloropyridines by arylmagnesium compounds.

## Acknowledgements

We thank the DFG (Leibniz-Program) and the CNRS for financial support. We thank Chemetall GmbH (Frankfurt) and the Degussa AG (Hanau) for the generous gift of chemicals.

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- Typical procedure, preparation of methyl 4-(5-cyano-2-pyridyl) benzoate (**3d**): Isopropylmagnesium chloride (2 M, 1.2 mmol in ether) was added to a solution of methyl 4-iodobenzoate (1.2 mmol) in THF (4 mL) at –40°C under argon and the mixture was stirred for 40 min leading to the functionalized arylmagnesium chloride. The resulting mixture was transferred into a solution of 2-bromo-5-cyanopyridine (**1c**) (1 mmol), bis(dibenzylideneacetone)palladium(0) (5.0 mol%) and 1,1'-bis(diphenylphosphino)ferrocene (5.0 mol%) in THF (2 mL) at –40°C. Stirring was continued at this temperature for 6 h followed by quenching using an aqueous saturated NH<sub>4</sub>Cl solution (5 mL). Extraction by diethyl ether, drying over MgSO<sub>4</sub> and solvent removal afforded a crude solid, which was purified by flash chromatography on silica (elution with CH<sub>2</sub>Cl<sub>2</sub>) affording the pure product **3d** as a white solid (mp = 162°C, 210 mg, 86% yield).
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